

## **Department of Food Technology**

**College of Agricultural Engineering Sciences** 

**University of Salahaddin** 

**Subject: Advanced Separation methods** 

**Course Book –** Postgraduate-PhD study

Lecturer's name: Assist Prof Dr Bashdar Abuzed Sadee

Academic Year: 2023/2024

## \*Course Book

| 1. Course name        | Autumn semester  |
|-----------------------|--|
| 2. Lecturer in charge | Assist Prof Dr Bashdar Abuzed Sadee                            |
| 3. Department/        | Food Technology Department / Agricultural Engineering Sciences |
| College               | College  |
| 4. Contact            | e-mail:bashdar.sadee@su.edu.krd                                |
|                       | Tel: 009647504681186   |
| 5. Time (in hours)    | Theory: 2  |
| per week              |  |
| 6. Office hours       | 5 day a week   |
| 7. Course code        |  |
| 8. Teacher's          | https://academics.su.edu.krd/bashdar.sadee                     |
| academic profile      | https://scholar.google.com/citations?user=a89t09EAAAAJ&hl=en   |
| 9. Keywords           | Separation techniques, HPLC, GSMS, Extraction, Solid phase     |
|                       | extraction, Liquid Chromatography, Precipitation               |

#### 10. Course overview:

In this section the lecturer shall write an overview about the subject he/she is giving. The course overview must cover:

Separations are extremely important in synthesis, in industrial chemistry, in the biomedical sciences, and in chemical analyses. Separation methods are techniques used to isolate and purify substances from a mixture. These methods are crucial in various scientific, industrial, and laboratory settings where the goal is to obtain pure components for analysis, use, or further processing. There are several types of separation methods, each suitable for specific applications. These separation methods can be used alone or in combination to achieve the desired level of purity and separation efficiency for a given application. The choice of method depends on the specific properties of the substances involved, the scale of the process, and the intended outcome. Separation is a preliminary step of analysis. The student will be introduced to the concept and principles of various separation methods.

- ■The importance of studying the subject
- Understanding of the fundamental concepts of the course
- Principles and theories of the course

will be free from interferences.

- A sound knowledge of the major areas of the subject
- Sufficient knowledge and understanding to secure employment This should not be less than 200 words

This course provides an introduction to the fundamental principles and theories of Separation techniques. It drives the researches how to isolate analyte of interest based on the actual types of the samples of different origin. It also, leads how to get pure analye that

#### 12. Student's obligation

Students must follow the instructions illustrated by the department. The students must attend all classes otherwise the lecturer should report the administration at the department.

Students must take quiz, midterm examination, writing an article review and deliver a presentation through the course.

#### 13. Forms of teaching

```
لنره ماموّستا رینگهی وانه ووتنهوه دهنووسیّت، بو نموونه: داتاشوّ و پاوهرپوّینت، سهر تهخته وهش، تهختهی سپی،
سمارتبوّردیان مهلزهمه... هتد
```

Different teaching approaches will be used in the class in order to deliver the main topics of the subject. Prior the class the hard copy of the lecture will be passed to the students in order to make notes by them. Different forms of teaching will be used to the reach the objectives of the subject such as using data show to deliver the presentation and other classical tools for more clarification.

#### 14. Assessment scheme

Breakdown of overall assessment and examination

لێره ماموّستا جوّری هملْسەنگاندن (تاقیکردنمومکان یان ئەزموونمکان) دەنووسێت بوٚ نموونه تاقیکردنمومی مانگانه، کویزهکان، بیرکردنمومی رمخنمگرانه (پریزهنتمیشن)، راپوّرت نووسین، ووتار نووسین یان ئامادەنمبوونی خوێندکار له یوّلدا…هند. ئامانه چهند نمرمی لمسمردهبیّت و ماموّستا چوّن نمرهکان دابهشدهکات؟

First Exam after five lectures

Second Exam after ten lectures.

Mark Distribution:

Midterm Exam, 20%

Quiz, 5%

Seminar, 10%

Article Review, 15%

Final Exam 50%

Final Mark 100%

#### 15. Student learning outcome:

```
پرکردنه وهی ئهم خانه یه زور گرنگه، ماموّستا دهرئه نجامه کانی فیّربوون دهنووسیّت. بو نموونه: روونی ئامانجه سهر هکیه کانی کوّرسه که (بابه ته که) بو خویّندکار
گونجاندنی ناوه روّکی کوّرسه که به پیّریستی دهرهوه و باز ار ی کار
قوتابی چی نوی فیّرده بیّت له ریّگه ی بیّدانی ئهم کوّرسه وه؟
```

This should not be less than 100 words

At the end of the course the students will learn how to separate interferences (unwanted substance) from the analyte of interest. They are also able to use various technique for purification of samples that required prior the analysis. Students will learn how to use this information in the next stage of their study.

Students should know the basic principles and have actual practice with the operational techniques of a wide variety of Analytical methods. In addition, they should be familiar with a great many other methods of separation that may be useful in the future.

### **16. Course Reading List and References:**

- Fundamentals of Analytical Chemistry; Eighth Edition, by Douglas A. Skoog, Donald M. West, F. James Holler and Stanley R. Crouch. by: Skoog& West
- 2. Dynamics of Chromatography: Principles and Theory by: J. Calvin Giddings
- 3. Techniques and Practice of Chromatography by: Raymond P.W. Scott

| 17. The Topics:   | Lecturer's name   |
|---|-------------------|
| Week 1: Introduction to separation Methods                  | Lecturer's name   |
| Week 2: Separation by Precipitation, Separations Based on   | Dr Bashdar Abuzed |
| Control of Acidity and Sulfide Separations                  | ex:(2 hrs)        |
| Week 3: Separations by Other Inorganic Precipitants ,       |                   |
| Separation of Species Present in Trace Amounts by           |                   |
| Precipitation and Salt-Induced Precipitation of Proteins,   |                   |
| Separation by Extraction                                    |                   |
| Week 4: Separating Ions by Ion Exchange and Applications of |                   |
| Ion-Exchange Methods  |                   |
| Week 5: Midterm Examination                                 |                   |
| Week 6: Chromatography                                      |                   |
| Definition ,-Chromatographic Theories and Classification    |                   |
| of Different Chromatographic Methods                        |                   |
| Week 7: Partition Chromatography                            |                   |
| -Mobile phase Operations in Chromatography                  |                   |
| 1- Elution  |                   |
| 2- Frontal  |                   |
| 3- Displacement   |                   |

Ministry of Higher Education and Scientific research Week 8: Paper Chromatography (PC) Type of paper Choice of solvent in PC Techniques of Paper Chromatography I-Ascending Paper Chromatography. II-Descending Paper Chromatography. III-Two-dimensional Separations on Paper IV-Horizontal or Circular Paper Chromatography Preparative paper chromatography Quantitative application of PC Week 9: Thin Layer Chromatography (TLC) - Additive to the Stationary layer in TLC Week 10: Ion Exchange Chromatography (IEC) -Classification of Ion-Exchanger -Resin Capacity Week 11: Liquid Chromatography (LC) Week 12: High performance Liquid Chromatography (HPLC)

-Elution System in HPLC

1- Isocratic Elution

2- Gradient Elution

Week 13: Ion-Pair Chromatography

Ion Chromatography

Week 14 Gas Chromatography

## 18. Practical Topics (If there is any)

#### 19. Examinations:

**1.** What are separation methods? Explain them

## Answer

| Separation Methods              |   |  |  |  |
|---------------------------------|---|--|--|--|
| Method                          | Basis of Method   |  |  |  |
| Mechanical phase separation     |   |  |  |  |
| a. Precipitation and filtration | Difference in solubility of compounds formed  |  |  |  |
| b. Distillation                 | Difference in volatility of compounds   |  |  |  |
| c. Extraction                   | Difference in solubility in two immiscible liquids  |  |  |  |
| d. Ion exchange                 | Difference in interaction of reactants with ion-exchange resin                                  |  |  |  |
| 2. Chromatography               | Difference in rate of movement of a solute through a stationary phase                           |  |  |  |
| 3. Electrophoresis              | Difference in migration rate of charged species in an electric field                            |  |  |  |
| 4. Field-flow fractionation     | Difference in interaction with a field or gradient applied perpendicular to transport direction |  |  |  |

# 2. What are class of column chromatographic methods Answer

| Classification of Column Chromatographic Methods |                      |                              |                           |  |  |
|--|----------------------|------------------------------|---------------------------|--|--|
| General Classification                           | Specific Method      | Stationary Phase             | Type of Equilibrium       |  |  |
| 1. Gas chromatography (GC)                       | a. Gas-liquid (GLC)  | Liquid adsorbed or bonded    | Partition between gas and |  |  |
|  |                      | to a solid surface           | liquid                    |  |  |
|  | b. Gas-solid         | Solid                        | Adsorption                |  |  |
| 2. Liquid Chromatography (LC)                    | a. Liquid-liquid, or | Liquid adsorbed or bonded    | Partition between         |  |  |
|  | partition            | to a solid surface           | immiscible liquids        |  |  |
|  | b. Liquid-solid, or  | Solid                        | Adsorption                |  |  |
|  | adsorption           |                              |                           |  |  |
|  | c. Ion exchange      | Ion-exchange resin           | Ion exchange              |  |  |
|  | d. Size exclusion    | Liquid in interstices of a   | Partition/sieving         |  |  |
|  |                      | polymeric solid              | _                         |  |  |
|  | e. Affinity          | Group specific liquid bonded | Partition between surface |  |  |
|  |                      | to a solid surface           | liquid and mobile liquid  |  |  |
| 3. Supercritical fluid                           |                      | Organic species bonded       | Partition between         |  |  |
| chromatography                                   |                      | to a solid surface           | supercritical fluid       |  |  |
| (SFC) (mobile phase:                             |                      |                              | and bonded surface        |  |  |
| supercritical fluid)                             |                      |                              |                           |  |  |

#### 20. Extra notes:

Here the lecturer shall write any note or comment that is not covered in this template and he/she wishes to enrich the course book with his/her valuable remarks.

## پيداچوونه وه هاوه ڵ 21. Peer review

This course book has to be reviewed and signed by a peer. The peer approves the contents of your course book by writing few sentences in this section.

(A peer is person who has enough knowledge about the subject you are teaching, he/she has to be a professor, assistant professor, a lecturer or an expert in the field of your subject).

ئهم كۆرسبووكه دەبێت لەلايەن هاوەڵێكى ئەكادىمىيەوە سەير بكرێت و ناوەڕۆكى بابەتەكانى كۆرسەكە پەسەند ُبكاتُ و جەند ووشەيەك بنووسێت لەسەر شياوى ناوەڕۆكى كۆرسەكە و واژووى لەسەر بكات. هاوەڵ ئەو كەسەيە كە زانيارى ھەبێت لەسەر كۆرسەكە و دەبيت پلەى زانستى لە مامۆستا كەمتر نەبێت.