

Ministry of Higher Education and Scientific research



Department of Food Technology

College of Agricultural Engineering Sciences

University of Salahaddin

Subject: Advanced Separation methods

Course Book – Postgraduate-PhD study

Lecturer's name: Assist Prof Dr Bashdar Abuzed Sadee

Academic Year: 2023/2024

***Course Book**

1. Course name	Autumn semester
2. Lecturer in charge	Assist Prof Dr Bashdar Abuzed Sadee
3. Department/ College	Food Technology Department / Agricultural Engineering Sciences College
4. Contact	e-mail: bashdar.sadee@su.edu.krd Tel: 009647504681186
5. Time (in hours) per week	Theory: 2
6. Office hours	5 day a week
7. Course code	--
8. Teacher's academic profile	https://academics.su.edu.krd/bashdar.sadee https://scholar.google.com/citations?user=a89t09EAAAAJ&hl=en
9. Keywords	Separation techniques, HPLC, GSMS, Extraction, Solid phase extraction, Liquid Chromatography, Precipitation
10. Course overview:	
<p>In this section the lecturer shall write an overview about the subject he/she is giving. The course overview must cover:</p> <p>Separations are extremely important in synthesis, in industrial chemistry, in the biomedical sciences, and in chemical analyses. Separation methods are techniques used to isolate and purify substances from a mixture. These methods are crucial in various scientific, industrial, and laboratory settings where the goal is to obtain pure components for analysis, use, or further processing. There are several types of separation methods, each suitable for specific applications. These separation methods can be used alone or in combination to achieve the desired level of purity and separation efficiency for a given application. The choice of method depends on the specific properties of the substances involved, the scale of the process, and the intended outcome. Separation is a preliminary step of analysis. The student will be introduced to the concept and principles of various separation methods.</p> <ul style="list-style-type: none"> ▪ The importance of studying the subject ▪ Understanding of the fundamental concepts of the course ▪ Principles and theories of the course ▪ A sound knowledge of the major areas of the subject ▪ Sufficient knowledge and understanding to secure employment <p>This should not be less than 200 words</p> <p>This course provides an introduction to the fundamental principles and theories of Separation techniques. It drives the researches how to isolate analyte of interest based on the actual types of the samples of different origin. It also, leads how to get pure analyte that will be free from interferences.</p>	

Students should know the basic principles and have actual practice with the operational techniques of a wide variety of Analytical methods. In addition, they should be familiar with a great many other methods of separation that may be useful in the future.

16. Course Reading List and References:

1. Fundamentals of Analytical Chemistry; Eighth Edition, by Douglas A. Skoog, Donald M. West, F. James Holler and Stanley R. Crouch. by: Skoog& West
2. Dynamics of Chromatography: Principles and Theory by: J. Calvin Giddings
3. Techniques and Practice of Chromatography by: Raymond P.W. Scott

17. The Topics:	Lecturer's name
<p>Week 1: Introduction to separation Methods</p> <p>Week 2: Separation by Precipitation, Separations Based on Control of Acidity and Sulfide Separations</p> <p>Week 3: Separations by Other Inorganic Precipitants , Separation of Species Present in Trace Amounts by Precipitation and Salt-Induced Precipitation of Proteins, Separation by Extraction</p> <p>Week 4: Separating Ions by Ion Exchange and Applications of Ion-Exchange Methods</p> <p>Week 5: Midterm Examination</p> <p>Week 6: Chromatography Definition , -Chromatographic Theories and Classification of Different Chromatographic Methods</p> <p>Week 7: Partition Chromatography -Mobile phase Operations in Chromatography</p> <ol style="list-style-type: none"> 1- Elution 2- Frontal 3- Displacement 	<p>Lecturer's name Dr Bashdar Abuzed</p> <p>ex:(2 hrs)</p>

<p>Week 8: Paper Chromatography (PC)</p> <p>Type of paper</p> <p>Choice of solvent in PC</p> <p>Techniques of Paper Chromatography</p> <p>I-Ascending Paper Chromatography.</p> <p>II-Descending Paper Chromatography.</p> <p>III-Two-dimensional Separations on Paper</p> <p>IV-Horizontal or Circular Paper Chromatography</p> <p>Preparative paper chromatography</p> <p>Quantitative application of PC</p> <p>Week 9: Thin Layer Chromatography (TLC) - Additive to the Stationary layer in TLC</p> <p>Week 10: Ion Exchange Chromatography (IEC)</p> <p>-Classification of Ion-Exchanger</p> <p>-Resin Capacity</p> <p>Week 11: Liquid Chromatography (LC)</p> <p>Week 12: High performance Liquid Chromatography (HPLC)</p> <p>-Elution System in HPLC</p> <p>1- Isocratic Elution</p> <p>2- Gradient Elution</p> <p>Week 13: Ion-Pair Chromatography</p> <p>- Ion Chromatography</p> <p>Week 14 Gas Chromatography</p>	
18. Practical Topics (If there is any)	

19. Examinations:

1. What are separation methods? Explain them

Answer

Separation Methods	
Method	Basis of Method
1. Mechanical phase separation	
a. Precipitation and filtration	Difference in solubility of compounds formed
b. Distillation	Difference in volatility of compounds
c. Extraction	Difference in solubility in two immiscible liquids
d. Ion exchange	Difference in interaction of reactants with ion-exchange resin
2. Chromatography	Difference in rate of movement of a solute through a stationary phase
3. Electrophoresis	Difference in migration rate of charged species in an electric field
4. Field-flow fractionation	Difference in interaction with a field or gradient applied perpendicular to transport direction

2. What are class of column chromatographic methods

Answer

Classification of Column Chromatographic Methods			
General Classification	Specific Method	Stationary Phase	Type of Equilibrium
1. Gas chromatography (GC)	a. Gas-liquid (GLC)	Liquid adsorbed or bonded to a solid surface	Partition between gas and liquid
	b. Gas-solid	Solid	Adsorption
2. Liquid Chromatography (LC)	a. Liquid-liquid, or partition	Liquid adsorbed or bonded to a solid surface	Partition between immiscible liquids
	b. Liquid-solid, or adsorption	Solid	Adsorption
	c. Ion exchange	Ion-exchange resin	Ion exchange
	d. Size exclusion	Liquid in interstices of a polymeric solid	Partition/sieving
	e. Affinity	Group specific liquid bonded to a solid surface	Partition between surface liquid and mobile liquid
3. Supercritical fluid chromatography (SFC) (mobile phase: supercritical fluid)		Organic species bonded to a solid surface	Partition between supercritical fluid and bonded surface

20. Extra notes:

Here the lecturer shall write any note or comment that is not covered in this template and he/she wishes to enrich the course book with his/her valuable remarks.

21. Peer review

پیداچوونہوی ہاوہل

This course book has to be reviewed and signed by a peer. The peer approves the contents of your course book by writing few sentences in this section.

(A peer is person who has enough knowledge about the subject you are teaching, he/she has to be a professor, assistant professor, a lecturer or an expert in the field of your subject).

نہم کورسبوو کہ دہبیت لہ لایہن ہاوملکی نیکادیمیہوہ سہیر بکرنیت و ناوہرؤکی بابہتہکانی کورسہکہ پھسند بکات و جہند ووشہیہک بنوسیت لہسہر شیاوی ناوہرؤکی کورسہکہ و واژووی لہسہر بکات. ہاوہل نہو کہسہیہ کہ زانیاری ہبیت لہسہر کورسہکہ و دہبیت پلہی زانستی لہ ماموستا کہمتر نہبیت.