

Department ofGo	eneralscience
College ofBasic E	ducation
University ofSala	haddin
Subject:Co Practical Inorganic	urse Book of Chemistry for Second Stage
Lecturer's name	Gulkhater Hammad Sharif

**Academic Year: 2022 - 2023** 

# **Course Book**

1. Course name	Practical Inorganic Chemistry	
2. Lecturer in charge	Gulkhater Hammad Sharif	
3. Department/ College	General Science /Basic Education	
4. Contact	e-mail: gulkhater.sharif@su.edu.krd	
	Tel: 07504750226	
5. Time (in hours) per week	12	
6. Office hours	30	
7. Course code		
8. Teacher's academic	1985-1986 BSc in Chemistry	
profile	1991 published a research in Journal of the Iraqi chemical	
	society ,subject of the research ( The stability constants	
	and thermodynamic functions of Palladium II (Pd II )	
	complexes with Salicylic acid its derivatives	
	1986-2001 assistant chemist	
	2001 – 2004 MSc in Analytical Chemistry	
	2004 till now Lecturer in General Science	
	Have been teaching these subjects:	
	<ul> <li>Analytical chemistry (theory lectures of second</li> </ul>	
	stage)	
	<ul> <li>Analytical chemistry ( practical of second stage)</li> </ul>	
	<ul> <li>Organic chemistry (theory lectures of third stage )</li> </ul>	
	<ul> <li>Industrial chemistry (theory lectures of third</li> </ul>	
	stage)	
	<ul> <li>Scientific debate (theory lectures of first stage)</li> </ul>	
	<ul> <li>General Chemistry of first stage (Theory and</li> </ul>	
	Practical)	
	<ul> <li>Had a training course about Methods of teaching,</li> </ul>	
	a course about PowerPoint, Excel, Microsoft	
	word, Moodle, Google Application.	
9. Keywords	Inorganic chemistry, purification, preparation,.	
40 0		

### 10. Course overview:

This course provides an introduction to the preparation of some inorganic compounds and procedures for separation of some iones .

It will teach the students how to do:

- 1- Separation of ions
- 2- Purification of crude table salt

#### 11. Course objective:

The purpose of this course to:

- Learn the student to prepare some compounds
- Learn the student procedure of purification
- Learn the student to separate some ions

#### 12. Student's obligation

- ❖ In each practical session a guiz will be done.
- ❖ Homework will be given to the students and bring it next session.
- The students should prepare a report after each practical session after doing any experiment.
- The students are not allowed to enter the laboratory without wearing the white coat.

## 13. Forms of teaching

The data show and whiteboard are used

#### 14. Assessment scheme

Semester	Practical	Theory		
	degree %	Seasonal Exam.	Quizzes	Dailey Activities
1 <sup>st</sup> semester	35			
2 <sup>nd</sup> semester	35			
Final Exam.				
<b>Total 100%</b>				

## 15. Student learning outcome:

- Through the practical sessions the students will learn the difference between tap water, mineral water and distilled water.
- Students will learn procedure for refining crude table salt
- Students learn information about Solubility

## 16. Course Reading List and References:

- Key references:
- 1.Basic inorganic chemistry, by F.Albert cotton
- 2.Inorganic chemistry by Catherine E. Housecroft
- 3. Coordination chemistry by Fred basolo
- 4. Scientific webs and journals

Useful references:		
17. The Topics:	Lecture	r's name
•		
18. Practical Topics (If there is any)		
1-Course book	Lecturer' Gulkhate Sharif (3 hrs)	s name: r Hammad
2-Refining Crude Table salt (NaCl)		
3- Separation of ions		
4-Determination of K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub> Solubil Water at Different Temperatures	lity in	
5- Preparation of Copper (I) Chloride (CuCl)	e	
6-Preparation of Copper (I) Iodide(C	CuI)	
7-Determination of Equivalent Weigl Zinc(Zn)	nt of	

8-Preparation of Chromium alum(Double salt ) KCr(SO <sub>4</sub> ) <sub>2</sub> .7H <sub>2</sub> O	
9- Hardness of Water	
10- Determination of Total Hardness of Water	
11- Determination of Permanent Hardness of Water	
19. Examinations:	
20. Extra notes:	

21. Peer review	پيداچوونهوهی هاوهڵ