



Department of Chemistry

College of Science

University of Salahaddin-Erbil

Subject: First semester-Organic Chemistry I
Course Book – stage 2

Lecturer's name:

Theoretical: Dr.Muslih S. Hamasharif .

**Practical: Dr. Muslih S. Hamasharif, Dr. Karzan kh. Hameed, Lecturer
Bushra H.Marbeen, Lecturer Shatha j. Dawood, Assiss. Lecture Shelan
H. Boya,**

Tuesday: 8:30 am – 9:30 am Group B Theoretical

10:30am- 11:30 am Group A Theoretical

Thursday: 8:30 am –10:30 am Group A Theoretical

10:30 am – 12:30 am Group B Theoretical

Sunday: 8:30 am- 5:30 pm Practical

Monday: 8:30 am– 5:30 pm Practical

Course Book

First semester

1. Course name	Practical Organic Chemistry
2. Lecturer in charge	(Theoretical) Dr.Muslih S. Hamasharif . (Practical) Dr Muslih S. Hamasharif, Dr.karzan kh. Hameed, Lecturer Bushra H. Marbeen, Lecturer Shatha J. Dawood, Assiss. Lecture Shelan H.Boya,
3. Department/ College	Chemistry/Science
4. Contact	muslih.hamashref1@su.edu.krd , karzan.hameed@su.edu.krd , Bushramarbeen@su.edu.krd , Shelanboya@su.edu.krd , shatha.dawood@su.edu.krd Tel: (00964750)
5. Time (in hours) per week	Theoretical: 3 hr/week Practical: 3 hr/week
6. Office hours	9am-1pm all over the week
7. Course code	
8. Teacher's academic profile	
9. Keywords	Organic compounds, structure and reactivity, physical properties, synthesis, reactions, identification
10. Course overview:	<p>Organic Chemistry is one of the core courses for chemistry students. Teaching this subject is accomplished through both theoretical and practical sessions. The theoretical part involves introduction to the basics of organic chemistry, in addition to different groups of organic compounds and the effect of the presence of various functional groups on physical and chemical properties of each group.</p> <p>The practical part is aimed at familiarizing the students with the basic skills required in organic chemistry laboratories.</p> <p>The course teaches the students experiments used in separation and identification of organic compounds in the first stage of this course. This includes learning the techniques for determination of physical properties of organic compounds such as melting and boiling points, methods used in purification of organic compounds based on their physical properties (e.g., sublimation, re-crystallisation and distillation) in addition to studying the solubility behavior of organic molecules and their applications in the laboratory and basic chemical tests used in qualitative elemental analysis of organic compounds. The second stage of this practical course includes experiments targeting different functional group reaction. the main purpose of the second part is to allow the student to learn the necessary</p>

lab skills associated with the topics that are taken in the theoretical class.

11. Course objective:

This course aims to familiarize the students with background knowledge and the basic skills required in all later stages of studying and working in organic chemistry-related fields. This is achieved through:

1. Introducing the students with complete and rich foundation of organic chemistry through studying different classes of organic compounds.
2. Enhance the students awareness and familiarize them with different classes of compounds, the associated difference in reactivity and strategies used in their synthesis.
3. Providing the students with knowledge of basic lab skills and the associated background theory.
4. Allowing the students to apply these skills in solving organic chemistry problems such as purification of compounds, isolation of different reaction products and their identification.
5. Familiarize the students with basic safety practices in an organic chemistry lab through studying hazards of different materials involved in the practical and measures of their control.

12. Student's obligation

- Students have to attend weekly practical sessions.
- Students will have to adhere to lab standards including attendance, fulfilling tasks and assignments and obliging to lab safety rules.
- Students will have to sit a minimum of two exams

13. Forms of teaching

Learning resources in this course include white board, lecture notes, PowerPoint presentations and media files.

14. Assessment scheme

Theoretical:

First semester exam: 8.5 marks

Second semester exam: 8.5 marks

Quiz: 6 marks

Attendance and class activities: 2 marks

Total is equal to 25 marks

Practical:

First semester practical Exam = 10 marks

Quiz = 5 marks

Reports and seminar = 3 marks

Student attendance and conduct in practical sessions (Activity)= 2 marks

Unknown = 5 Marks

Total is equal to 25 marks

15. Student learning outcome:

The principal learning outcome of this course is

- to build the background knowledge required at all later levels of organic chemistry
- to help the student grasp the theoretical understanding of the course and to demonstrate materials taught in lecture and promote interest in organic chemistry
- to familiarize the student with skills and materials used in organic chemistry laboratories

16. Course Reading List and References:

- (1) Morrison, R. T., and R. N. Boyd. *Organic Chemistry*, 6th Edition.
- (2) *Organic Chemistry*, sixth edition, John McMurry.
- (3) *Organic Chemistry*, Ninth edition, Solomons and Fryhle.
- (4) *Vogel's Textbook of Practical Organic Chemistry*, B.S. Furniss and others, Longman Group Limited London.
- (5) *The Systematic Identification of Organic Compounds*, Ralph L. Shriner and others, John Wiley and Sons, USA.
- (6) *Practical Organic Chemistry*, Frederick G. Mann and Bernard C. Saunders, Longman Group Limited London.
- (7) *Organic Experiments*, Louis F. Fieser and Kenneth L. Williamson, D. C. Heath and company.
- (8) *Laboratory Manual of Organic Chemistry*, Raj K. Bansal, New age international (p) limited, Publishers

17. The Topics:

Lecturer's name

Theoretical

First semester: Organic Chemistry I

Week 1 (Introduction to Organic Chemistry)

Definition of organic chemistry. Chemical bonds (covalent, ionic bonds and Hydrogen bonds). Electronic configuration. Polarity of bonds
Electronic effects, electronegativity, inductive effects, resonance.
Physical Forces.

Week 2-4 (Aromatic Compounds and Electrophilic Substitution Reaction)

Aromatic hydrocarbons (benzene), Aromaticity , Orbital Picture, Stability of benzene ring, Aromatic character (Huckel rule)
Nomenclature , Electrophilic aromatic substitution, Nitration , Sulphonation , Friedel-Craft Reactions , Halogenation, Effect of substitution group, Activity and deactivating groups, Reactivity and orientation, Arenes (Aryl halides), Nomenclature, Physical Properties,

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<p>Preparation and rearrangement of carbonium ion , Reactions, Problems.</p> <p>Week 5-6 (Alkyl halide) Definition, physical properties, Nomenclature Preparation, Reactions, Nucleophilic substitution reaction, SN1 Mechanism, SN2 Mechanism, Elimination reaction, E1 mechanism, E2 mechanism, Problems.</p> <p>Week 7-9 (Alcohols) Alcohols , Nomenclature, Physical Properties, Preparation , Reaction,</p> <p>Week 10-11 (Phenols) Phenols , Structure, Nomenclature, Physical Properties, Salts of Phenol, Preparation, Reaction, Acidity, Ester Formation, Ring substitution.</p>	
<p>18. Practical Topics (If there is any)</p>	
<ol style="list-style-type: none"> 1. Introduction to organic chemistry lab., general safety principles and instructions 2. Solubility behaviour of organic compounds: water soluble compounds 3. Solubility behaviour of organic compounds: water insoluble compounds 4. Elemental analysis of organic compounds: fusion with sodium 5. Extraction salt from oil 6. Acid-Base Extraction- based on acidity, basicity and Solubility 7. Application of Acid- Base Extraction 8. Elemental analysis of organic compounds: fusion with sodium 9. Functional Group 	<p>Dr.Muslih S. Hamasharif, Lect. BushraH.Marbeen, lect. ShelanH.Boya, Lect. peshawa shafiq Osw</p>

19. Examinations:

A typical exam question may include a combination of the following:

- Definitions
- Identifying the products of chemical reactions
- Giving explanations for facts and phenomena
- Outlining reaction mechanisms
- Suggesting solutions to problems encountered in practical organic chemistry (e.g., separation of a compound from a mixture)

20. Extra notes:

21. Peer review

پیداچونہوہی ہاوہل

نہم کورسبوو کہ دہبیت لہ لایمن ہاوہل کی ٹہکادیمیہوہ سہیر بکریت و ناوہرؤکی بابہتہکانی کورسہکہ پہسہند بکات و جہند ووشہیہک بنووسیت لہسہر شیاوی ناوہرؤکی کورسہکہ و واژووی لہسہر بکات۔
ہاوہل نہو کہسہیہ کہ زانیاری ہہبیت لہسہر کورسہکہ و دہبیت پلہی زانستی لہ ماموستا کہمتر نہبیت۔