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**Department of Chemistry**

**College of Science**

**Salahaddin University**

# Subject: Principles of Analytical Separation

**Theoretical Part**

**Course Book –3rd Year -1st semester**

**Lecturer's name: Dr. Rebwar Omar Hassan**

**Academic Year: 2022-2023**

**Course Book (Theory)**

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| **1. Course name** | **Principles of Analytical Separation** |
| **2. Lecturer in charge** | **Dr. Rebwar Omar Hassan** |
| **3. Department/ College** | **Chemistry - College of Science** |
| **4. Contact** | **e-mail:** **Rebwar.hassan@su.edu.krd** |
| **5. Time (in hours) per week**  | **Theory: 2**  |
| **6. Office hours** |  |
| **7. Course code** |  |
| **8. Teacher's academic profile** Academic achievements and Qualifications: (starting from the most recent degree)

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| From- To | Degree | College-University | Country |
| 2009 to date | PhD in Analytical Chemistry, Department of Chemistry | College of Science- University of Salahaddin | Iraq |
| 2001-2002 | M. Sc. in Analytical Chemistry, Department of Chemistry | College of Science- University of Salahaddin | Iraq |
| 1993-1997 | B.Sc. Chemistry, Department of Chemistry,  | College of Science- University of Salahaddin | Iraq |

1. ***Teaching Activities***

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| From- To | Subject | Stage-College | University |
| 20012- to date | I introduction to Analytical Chemistry-Volumetric Analysis | 1st- year students / Chemistry Department-College of Science | Salahaddin University |
| 2012 to date | Advance Separation Techniques | M.Sc student –Chemistry Department –College of Science | Salahaddin University |
| 2013- to date | Organic Functional Group Analysis | M.Sc student –Chemistry Department –College of Science | Salahaddin University |
| 2009-2011 | Analytical Chemistry- Separation Methods | 3rd- year students / Chemistry Department-College of Science | Salahaddin University |
| 2009 to date | Environmental Application of Analytical Chemistry  | 1st- year students / Environmental-Department-College of Science | Salahaddin University |
| 2003-2005 | Analytical Chemistry- Separation Methods-Practical Fiel | College of Science, University of Salahaddin | Salahaddin University |

**Published articles, papers and researches:*** + - 1. Separation and determination of Sulfamerazine and Sulfathiazole in Veterinary Drugs by Ion-Pair HPLC. (Tikrit Journal for Pure Science Vol.(9) No.(2) 2003).
			2. Spectrophotometric Determination of Vitamin B1 (Thiamin Hydrochloride) In Pharmaceutical Preparation by Coupling Reaction with Diazotized Sulfanilic acid.) Tikrit Journal of Pharmaceutical Science, Vol. (1), No (2), 2005).
			3. Separation and Determination of Five Water-Soluble Vitamins in Pharmaceutical Preparation by IP RP-HPLC.) Tikrit Journal of Pharmaceutical Science, Vol. (1), No (1), 2005 (.
			4. Determination of Contents Levels of Nitrogen Species (Nitrite, Nitrate and Nitrosamine) In Processed Meat Consumed In Erbil City. (Der Pharma Chemica, Vol. 2,No. 6, 2010, 31).
			5. Reverse-flow-injection analysis (FIA) for the determination of vitamin C in pharmaceutical formulation with chemiluminescence detection. (African Journal of Pure and Applied Chemistry Vol. 5, No. (11), 2011, 373).
			6. Reverse-FIA with Spectrophotometric Detection Method for Determination of Vitamin C. (J. Iran. Chem. Soc., Vol. (8), No.(3), 2011, 662).
			7. Indirect Spectrophotometric determination of Cephalexin in pharmaceutical formulations. (Chemical Science Transection, Vol.(2), No (4), 2013, 1110).
			8. Determination of Nicotinamide by Stopped-Flow Injection method in Pharmaceutical Formulations. (Arabian Journal of Chemistry, Vol. (6), 2013, 393).
			9. Reversed Flow injection method for the Spectrophotometric Determination of Cephalexin in Pharmaceutical Products. (Zanco Journal of pure and Applied Chemistry,Vol.(25), No. (3), 2013).
			10. Spectrophotometric determination of chlorthalidone in pharmaceutical formulations using different order derivative methods. (Arabian Journal of Chemistry, Under press)
			11. Reverse-FIA Technique for the Determination of Omeprazole Using Chemiluminescence Detection. (Zanco Journal of pure and Applied Chemistry, Under press)
			12. Simultaneous spectrophotometric determination of thiamine and pyridoxine in multivitamin dosage forms using H-point standard addition and Vierodt᾿s methods. (Journal of the Iranian Chemical Society, 15(7)(2018), 1603–1612)

***Conferences and Scientific activities:*** 1. Training courses about application international quality according to (*ISO/ IEC 17025:1999*) in analytical laboratory and quality control laboratory.
2. 1st International Conference of Chemical, Environment and Energy; ICCEE-2012, Malaysia.
3. Chemical Safety and Security Officer Training, 18th -20th June, 2013, University of Salahaddin, Iraq (Sponsored by the United State Department of State).
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| **9. Keywords** | Separation techniques, Extraction, Solid phase extraction, Distillation, Precipitation, Volatilization |
| **10. Course overview:** “Separation Science”, is a senior level of undergraduate course within the analytical chemistry lecture series. It is concerned with all aspects of the chemical and physical separation of substances, with particular emphasis on separation as a tool for chemical identification and measurement.Throughout this course, lecture topics will be illustrated with examples from biological, pharmaceutical, food & drink, environmental sample analysis. This will give a sense of the scope and impact of separation science on society, as well as providing concrete examples on how to identify appropriate separation and analysis strategies depending on the target analyte(s) and sample matrix. |
| **11. Course objective:**This course provides an introduction to the fundamental principles of Separation techniques. It will teach you how to correctly handle and interpret experimental measurements; you will also learn how to perform an analytical procedure. |
| **12. Student's obligation**Each student at the end of the course must be preparing a report about any separation methods other than that mentioned or discussed during the course. This report includes Theory, principle and discussion on the selected technique how it helps to improve the understanding of the principles. |
| **13. Forms of teaching**Data show and white board  |
| **14. Assessment scheme**The students are required to do two closed exams during the course period. All exams have 8% marks; the quiz tests have 5% marks, the attendance, classroom, activities, absence count 2% marks. So that the final grade will be based upon the following criteria:Exams (closed and optional): 8%Quiz tests and homework: 5%Classroom participation and assignments: 2%Final Exam: 50% ‌ |
| **15. Student learning outcome:**Students should know the basic principles and have actual practice with the operational techniques of a wide variety of Analytical methods. In addition, they should be familiar with a great many other methods of separation that may be useful in the future.  |
| **16. Course Reading List and References‌:**The student can find additional information and examples in the following references1. Modern Analytical Chemistry; by David Harvey.2. Fundamentals of Analytical Chemistry; Eighth Edition, by Douglas A. Skoog, DonaldM. West, F. James Holler and Stanley R. Crouch.3. Principles and Practice of Analytical Chemistry, Fifth Edition, by F.W. Fifield and D.Kealey.4. Vogels, Textbook of Quantitative Chemical Analysis, Fifth Edition, G.H. Jeffery, J.Bassett, J. Mendham and R.C. Denney.5. Quantitative Chemical analysis, Seventh Edition, -Daniel C. Harris. |
| **17. The Topics:** | **Lecturer's name** |
| Course Program (Principles of Analytical Separation)**-**General Introduction to Separation science1- IntroductionImportant total analysis process steps:* Sampling
* Methods
* Separation
* Quantitation
* Evaluation
	1. Separation Science

-Definition of Separation Science-Classification and Application of SeparationsI- Separation by Phase-Formation methods* Volatilization Methods (conversion of Solid to Gas)
* Distillation Methods (conversion of Liquid to Gas)
* Separation by Precipitation

II- Separation by Phase- transformation methods* Extraction
1. Solvent extraction (Liquid Extraction)
	1. The equilibrium constant
	2. Efficiency of extraction
	3. Selectivity of extraction
	4. Nature of partition forces
		1. Dispersion interaction
		2. Dipole-Dipole interaction
		3. Induction interaction
		4. 4-Hydrogen bonding

1.5. Extraction system1.5.1. Extraction of Covalent, Neutral Molecules1.5.1.1. pH effect1.5.1.2. Effect of complex formation1.5.1.3. Effect of association1.6. Methods of Extraction 1.6.1. Batch extraction1.6.2. Continuous extraction1.6.3. Multistage extraction1. Solid-Phase Extraction (Liquid-Solid Extraction)
 | Lecturer's name: Dr Rebwar Omar Hassan. |
| **20. Peer review پێداچوونه‌وه‌ی هاوه‌ڵ**   |