## Nuclear Physics /3<sup>rd</sup> Communication /2<sup>nd</sup> course

Q1/Determine the atomic mass and the mass excess of  $\binom{27}{13}Al$ .

Q2/Calculate the mass defect for  ${}_3^7$ Li. The mass of  ${}_3^7$ Li is 7.016003 amu.

Q3/ Determine the radii of a  $^{16}$ O and a  $^{208}$ Pb nucleus.

Q4/ If the radius of a nucleus is given by  $R=Ro A^{1/3}$  with Ro =1.2F, what is the density of the nuclear matter (a) in g/cm3, (b) in nucleons/F<sup>3</sup>.

Q5/ Determine the approximate density of a nucleus, if the nucleus is treated as a uniform from sphere.

Q6/The actual atomic mass of  ${}^{40}_{20}Ca$  is 39.96259. find the binding energy of this nuclide, using 1.008665 amu for the mass of a neutron and 1.007825 amu for the mass of atomic hydrogen. also calculate the binding energy per nucleon.

Q7/Determine the total binding energy and the average binding energy of the element  ${}_{3}^{7}$ Li, if you know that each of quantities are in the amu, where M( ${}_{3}^{7}$ Li)=7.016004 amu, Mn= 1.008665 amu MH=1.007825.

Q8/Two isotopes of oxygen  ${}^{16}_{8}O$  and  ${}^{18}_{8}O$  having the atomic masses (15.990523 u) and (17.994768 u) respectively. What do you expect the relative abundance of two isotopes.

Q9/ According to single particle model (shell model), what is the spin and parity of the ground state of  $_{19}^{39}$  K nucleus.

Q10/The atomic mass of Zinc isotope  $\binom{64}{30}Zn$  is (63.929 u). Compare binding energy between classical and liquid empirical mass formula.

(mp=1.007825u, mn=1.007825u, av=14.1, as=13, ac=0.595, aa=19, ap=33.5)

Q11/ Determine the ground state angular momentum of

 $^{16}_{\phantom{1}8}0,\ ^{15}_{\phantom{1}8}0\ ,\ ^{20}_{\phantom{1}10}N\ ,\ \ ^{14}_{\phantom{1}7}N$ 

Q12/ what will be the mass of a(10 curi) sample of (Co-60) given that its half life is (5.26 year).

Q13/ A certain radioactive substance has a decay constant (1.44  $\times$ 10<sup>-3</sup>yr), in what time will (75%) of the initial number of atoms disintegration.

Q14/ What is activity of one gram (Ra-226) whose half live is 1622 year.

Q15/ What is the mass sample of C-14 (t1/2 = 5570yr) that has on activity of 5 Ci?

Q16/ What is the activity of  $5 \times 10^{-7} Kg$  of (U-230) whose half life is (0.18  $\times 10^{4}$ Sec)?

Q17/ How much time required for amount of Sr-19 (t=28yr) to be reduced by 75%?