Kurdistan Region
Salahaddin University-Erbil
College of Engineering
Chemical-Petrochemical Engineering Department



Oxidation Desulfurization of Naphtha

A Project Submitted to the Chemical-Petrochemical Engineering Department

University of Salahaddin-Erbil

in the Partial Fulfillment of the Requirement for the Degree of Bachelor of Science

in Chemical-Petrochemical Engineering

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2023-2024

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Chapter 1 Introduction

Chapter one: 1 Introduction

Naphtha is a key intermediate product in the refining industry, obtained from the distillation of crude oil. It is a versatile hydrocarbon mixture consisting of various carbon chain lengths, typically ranging from C5to C12. Naphtha serves as a vital feedstock for numerous downstream processes, including petrochemical production and gasoline blending.

naphtha often contains sulfur compounds such as mercaptans, sulfides, disulfides, thiophene (Th), benzothiophene (BT), dibenzo thiophene (DBT), 4,6-dimethyl dibenzothiophene (4,6 DMDBT), benzonaphthothiophene (BNT), and so on. (Marsh, Hill et al. 2000)

These sulfur compounds can have detrimental effects on both the downstream processes and the environment. For instance, during the combustion of sulfur-containing fuels, sulfur compounds oxidize to form sulfur dioxide (SO2), a major contributor to air pollution and acid rain, respiratory diseases, cancer. Additionally, Sulfur compounds can act as catalyst poisons, particularly in

refining processes. They can deactivate or reduce the effectiveness of catalysts used in hydro-processing units, such as those for hydrocracking and hydrotreating processes, leading to reduced process efficiency and increased maintenance cost.

Hydrodesulfurization (HDS), have been widely used in refineries to reduce sulfur content. However, these methods have limitations in terms of their efficiency, cost, and environmental impact. Oxidative desulfurization (ODS) has emerged as a promising alternative technique to overcome these limitations. (Almaliky and Alkazrajy 2023).

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