Ministry of Higher Education and Scientific research



Department of ... General Science.

College ofBasic Education .

University ofSalahaddin

Subject:Biochemistry.

Course Book : (3- stage)

Lecturer's name. MSc. Snowber M. Ahmed

Academic Year (2021–2022)

1. Course name	Biochemistry		
2. Lecturer in charge	Lecturer. Snowber M. Ahmed		
3. Department/ College	General Science / Basic education collage		
4. Contact	e-mail: <u>snowber.ahmed@su.edu.krd</u>		
	Tel: 0754026887		
5. Time (in hours) per	Theory: 2		
week	Practical: 3		
6. Office hours	10 hrs		
7. Course code	6511		
8. Teacher's academic	In 1997 graduated from chemistry department in		
profile	Education college / Salahaddin University ,in 2004		
	I started worked in Koya Technical Instituted as		
	assist lecturer after qualify master's degree in		
	Biochemistry from Baghdad University Ibn-		
	Alhaitham college, in 2009 I was admission to		
	academic staff of general science department from		
	basic education college.		

Course Book

9. Keywords : bimolecular , carbohydrate ,lipid ,protein ,enzyme, nucleic acid

10. Course overview:

- Biochemistry is the study of the chemical processes of living things. we will cover the structures and functions of major biomolecules: chemical and physical properties of proteins, carbohydrates, and lipids; enzyme kinetics and mechanisms ; relationships of organ systems. At the end of the course, students will understand how the chemical and physical properties of bio molecules influence their function. Furthermore, they will be able to use this knowledge to describe how chemical changes alter the function of biological systems. The student will master new vocabulary and demonstrate an understanding of the molecular structure and function of biological molecules.

11. Course objective:

At the end of this course students should be able to:

- understand the important of biomolecules.
- Relationship between biochemistry and other science and another branches of chemistry .
- understand the basic molecular properties of 3 of the 4 classes of biological molecules (proteins, carbohydrates, and lipids....) and the subunits from which they are formed.
- explain how the structure of biological molecules dictates function and how changes in structure direct biochemical reactions.
- describe the catalytic functions of enzymes, and perform enzyme kinetics calculations.
- describe the network of chemical reactions that make up central metabolism.
- read and understand scientific literature pertaining to subject matter in biochemistry.
- apply knowledge & concepts to novel problems

12. Student's obligation

The role of students and their obligation throughout the academic year are the Attendance and completion of all tests, examinations, and homework's.

13. Forms of teaching

- white board
- hand out
- data show

14. Assessment scheme :- there is two main exams

- 1- May
 - Theory
 - Practical
- 2- Final exam

Total marks :

بەر يو بەر ايەتى دڭنيايى جۆرى و متمانەبەخشىن Directorate of Quality Assurance and Accreditation

(10 marks)

(15 marks)

(25 marks)

(50%)

15. Student learning outcome:

- Identify the levels of structure in carbohydrate and describe the stabilization of these structures.
- Identify the levels of structure in proteins and describe the stabilization of these structures.
- Describe the structure and mechanism of representative enzymes in biochemical pathways. Describe representative mechanisms of enzyme catalysis
- Describe the important of to the following molecular classes:
 - a. Carbohydrates
 - b. Fats and lipids
 - c. Amino Acids
 - d. Steroids
- Read and interpret scientific articles in biochemistry.

16. Course Reading List and References:

- 1- Trudy,M.,James,R.,(2003),Biochemistry the molecular basis of life (third edition).
- 2- J.I. Jain ,Sunjay ,J., and Nitin ,J.,(2005),Fundamentals of biochemistry ,Vol.(1)
- 3- Scientific webs and journals(Internet).
- 4- Fundamentals of Chemistry by David E. Goldberg

17. The Topics:	Lecturer's name
Introduction biochemistry, important, relationship between biochemistry and other s (2)	Snowber M. Ahmed sciences (hrs)
➢ Basic bio molecules in organism. (2)	2 hrs)

Ministry of Higher Education and Scientific research					
Carbohydrate :- definition ,classification ,					
important . (2 hrs)					
Proteins :- definition , function , classification					
Amino saida nontidos (Abra)					
Amino acids , peptides . (4hrs)					
Lipids :- definition, function, Important of					
lipids, classification. (4hrs)					
Nucleic acids :- definition, function, RNA,					
DNA					
Enzymes :- definition, general properties					
Enzymes classification, factors effecting on					
enzyme activity.					
chzynic activity.					
Hormones:- definition, function					
\succ Glands and their hormones ,					
Insulin ,adrenalin hormones					
18. Practical Topics (If there is any)					
	Dr. Bakhtiar Kakel				
Carbohydrate tests :-	Snowber M. Ahmed				
1- Molish test					
2- Benedicts test	(3 hrs)				
3- Barfoid test					
4- Bails test					
5- Sillivanofs test					
6- Oxidation of galactose					
7- Reaction of urea					
8- Starch hydrolysis					

 Protein test Burit test Burit test Xanthprotic test Lipid test Vitamin test 19. Examinations: Compositional: Explain how carbohydrate is produced ? with equation . What's difference between glycogen and cellulose . Enumerate the factors affecting on enzyme activity . Explain the important of thyroid gland . -Compare between RNA and DNA structure . -Compare between RNA and DNA structure . - Enumerate the type of lipoprotein in blood . 2-True or false type of exams: Q/Correct the following sentences An example of protein with no quaternary (4°) structure is DNA . Conjugated protein contains only amino acids. Oils are esters of fatty acids with higher molecular weight of alcohol. 3. Multiple choices: Q/Choose the best answer for each of the following:- 1-The structure unit of living systems is a-microorganism b-cell c- biomolecules d- carbohydrate 2-Galactose is a sugar. 	Ministry of Higher Educatio	on and Scientific research			
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•				d- carbohydrate	
a-aldopentose b- ketotetrose c-aldohexose d- ketohexose 3-In sphingo phospholipids the alchol is	a-aldopentose	b- ketotetrose		d- ketohexose	
a-glycerol b- sphingomyelin c- sphingosine d- isoprenoids		-		e d- isoprenoids	

21. Peer review

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